

# Influence of Cr-substitution on the magnetic properties of $\text{Fe}_{1-x}\text{Cr}_x\text{SnSbO}_6$

T. Groń<sup>1</sup>, E. Filipek<sup>2</sup>, G. Dąbrowska<sup>2</sup>, H. Duda<sup>1</sup> and M. Oboz<sup>1</sup>

<sup>1</sup>University of Silesia, Institute of Physics, ul. Uniwersytecka 4, 40-007 Katowice, Poland

<sup>2</sup>West Pomeranian University of Technology, Szczecin, Department of Inorganic and Analytical Chemistry, Al. Piastów 42, 71-065 Szczecin, Poland

The oxides  $\text{SnO}_2$ ,  $\text{Fe}_2\text{O}_3$ ,  $\text{Cr}_2\text{O}_3$  and  $\text{Sb}_2\text{O}_4$  and known compounds formed with the participation of these oxides, due to their chemical, magnetic, electrical and catalytic properties are particularly attractive for basic research and for a large number of prospective applications. A new continuous substitution solid solution of a  $\text{Fe}_{1-x}\text{Cr}_x\text{SnSbO}_6$  type is formed in the  $\text{FeSnSbO}_6$ – $\text{CrSnSbO}_6$  system and crystallize in the tetragonal system and they have the rutile-type structure [1,2]. The ultraviolet-visible and near-infrared measurements showed that the energy gap,  $E_g$ , in  $\text{Fe}_{1-x}\text{Cr}_x\text{SnSbO}_6$  increases monotonically with increasing content  $x$ , *i.e.*  $E_g = 1.67, 1.88, 1.89, 1.96$  and  $2.01$  eV for  $x = 0.0, 0.25, 0.5, 0.75$  and  $1.0$ , respectively [3]. Electrical measurements of  $\text{Fe}_{1-x}\text{Cr}_x\text{SnSbO}_6$  solid solution showed semiconducting behaviour with the activation energy decreasing from  $E_A = 0.64$  eV for  $x = 0.0$  to  $E_A = 0.32$  eV for  $x = 1.0$  in the intrinsic conductivity temperature region as well as the  $n$ -type conduction at room temperature. The  $I$ - $V$  characteristics and the conductance  $G$  at 300 and 400 K showed symmetrical and non-linear behavior in the voltage range (-100, 100 V) suggesting the electron emission over the potential barrier especially for the boundary compounds  $\text{FeSnSbO}_6$  and  $\text{CrSnSbO}_6$  [4].

Magnetization measurements were carried out using a Quantum Design System (MPMS XL). Static (dc) magnetic susceptibility was measured in the magnetic field  $H_{dc} = 1000$  Oe and recorded both in zero-field-cooled (ZFC) and field-cooled (FC) mode. Dynamic (ac) magnetic susceptibility was measured at an internal oscillating magnetic field  $H_{ac} = 3.9$  Oe with an internal frequency  $f = 300$  Hz. Both dc and ac magnetic susceptibility were measured in the temperature range 2–300 K. Magnetization isotherms were measured in the temperature range 2–300 K in static (dc) magnetic fields up to 70 kOe.

Magnetic measurements showed that the  $\text{Fe}_{1-x}\text{Cr}_x\text{SnSbO}_6$  solid solution is ferrimagnetic over the range of concentrations of chromium ions from  $x = 0.0$  to  $x = 1.0$ . Only for  $x = 1.0$  the long-range ferrimagnetic interaction with the Curie temperature  $T_C = 5$  K is observed. In all samples, a short-range antiferromagnetic interaction occurred, seen in a negative paramagnetic Curie-Weiss temperature that varied from -41 K for  $x = 0.0$  via -157 K for  $x = 0.5$  to -190 K for  $x = 1.0$ . The effective magnetic moment increased from  $3.585 \mu_B/\text{f.u.}$  for  $x = 0.0$  to  $4.99 \mu_B/\text{f.u.}$  for  $x = 1.0$ , suggesting the existence of both a mixed valence chromium and iron ions. The imaginary component of ac magnetic susceptibility showed the oscillating values close to zero, indicating a lack of the energy dissipation characteristic for the ferrimagnetic state and spin frustration. The most interesting observation was the appearance of the spin-glass-state with an increase in the content of chromium ions, for which the freezing temperature reached the highest value of 25 K for  $x = 1.0$ .

[1] J. Isasi, L. Veiga, C. Pico, *J. Mater. Sci. Lett.* **15**, 1022 (1996).

[2] V.A. Govorov, A.M. Abakumov, M.G. Rozova, A.G. Borzenko, S.Yu. Vassiliev, V.M. Mazin, M.I. Afanasov, P.B. Fabritchnyi, G.A. Tsirlina, E.V. Antipov, E.N. Morozova, A.A. Gippius, V.V. Ivanov, and G. Van Tendeloo, *Chem. Mater.* **17**, 3004 (2005).

[3] G. Dąbrowska, and E. Filipek, M. Piz, *Ceram. Int.* **41**, 12560 (2015).

[4] T. Groń, G. Dąbrowska, E. Filipek, H. Duda and B. Sawicki, *Acta Phys. Pol. A* **129**, 153 (2016).